

# Carbon Capture, Use and Storage/Sequestration - CCUS

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**PASSION FOR CHEMISTRY**

# The Problem

- Combustion of fossil fuels and biomass or the production of steel and cement generates CO<sub>2</sub> and H<sub>2</sub>O + Heat which is used to generate steam in thermal fossil fuel power stations or for transport. In a power station, the steam turns a turbine and electricity is generated.
- The Equation is:  $C_xH_yO_z + O_2 \rightarrow CO_2 + H_2O + \text{Heat} (+ N_2 + SO_x)$
- CO<sub>2</sub> is a powerful green house gas
- When air is used for combustion, N<sub>2</sub> and NO<sub>x</sub> are also produced in the flue gas.
- If coal that has sulphur in it is used, SO<sub>x</sub> is also produced

# The Solution

- Separate the  $N_2$  from the  $CO_2$  in the flue gas and capture/store/use the  $CO_2$ . This is known as Post-Combustion CCS.
- Separate  $O_2$  from  $N_2$  in ambient air and use pure  $O_2$  for the combustion reaction then only  $CO_2$  and  $H_2O$  is produced. This is called Oxy-Combustion.
- $CO_2$  can be stored in geological formations
- $CO_2$  can be used in Enhanced Oil Recovery
- $CO_2$  can be used as a Carbon source monomer for synthesizing larger molecules
- Reacting  $CO_2$  with  $NaOH$  produced Soda Ash or Sodium Carbonate

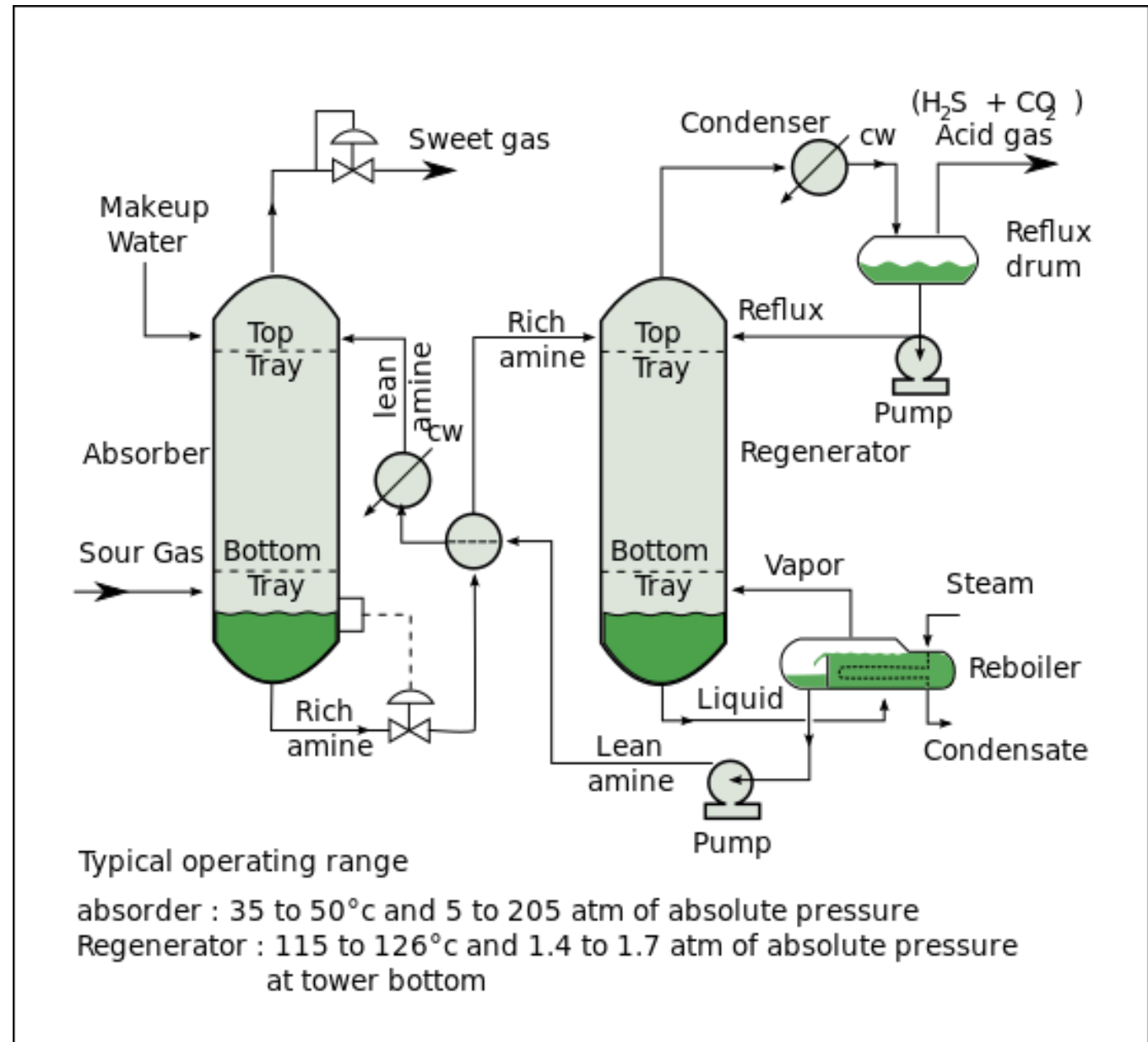
# The Chemistry

- Amine gas treating, also known as amine scrubbing, gas sweetening and acid gas removal, refers to a group of processes that use aqueous solutions of various alkylamines (commonly referred to simply as amines) to remove hydrogen sulfide (H<sub>2</sub>S) and/or carbon dioxide (CO<sub>2</sub>) from combustion flue gases.
- It is also possible to combust the fossil fuel incompletely to produce CO and H<sub>2</sub> called Syngas.
- Calcium looping processes for capturing CO<sub>2</sub> from large emissions sources are based on the use of CaO (lime) particles as sorbent in circulating fluidized bed reactors.
- Gas separating membranes are also being proposed that can separate CO<sub>2</sub> from N<sub>2</sub> and even recover H<sub>2</sub>O.

# Process Diagram of Amine Absorber >

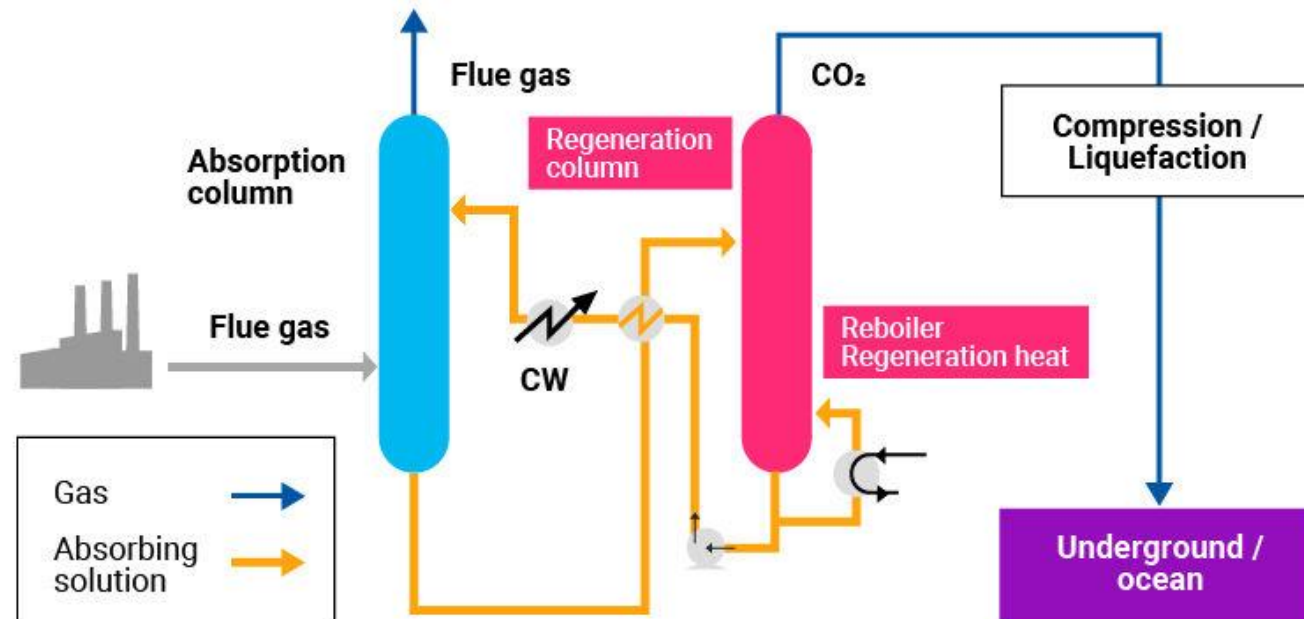
Animation of CCS from GE Power Below:

<https://www.youtube.com/watch?v=J8IRLH1nuCg>



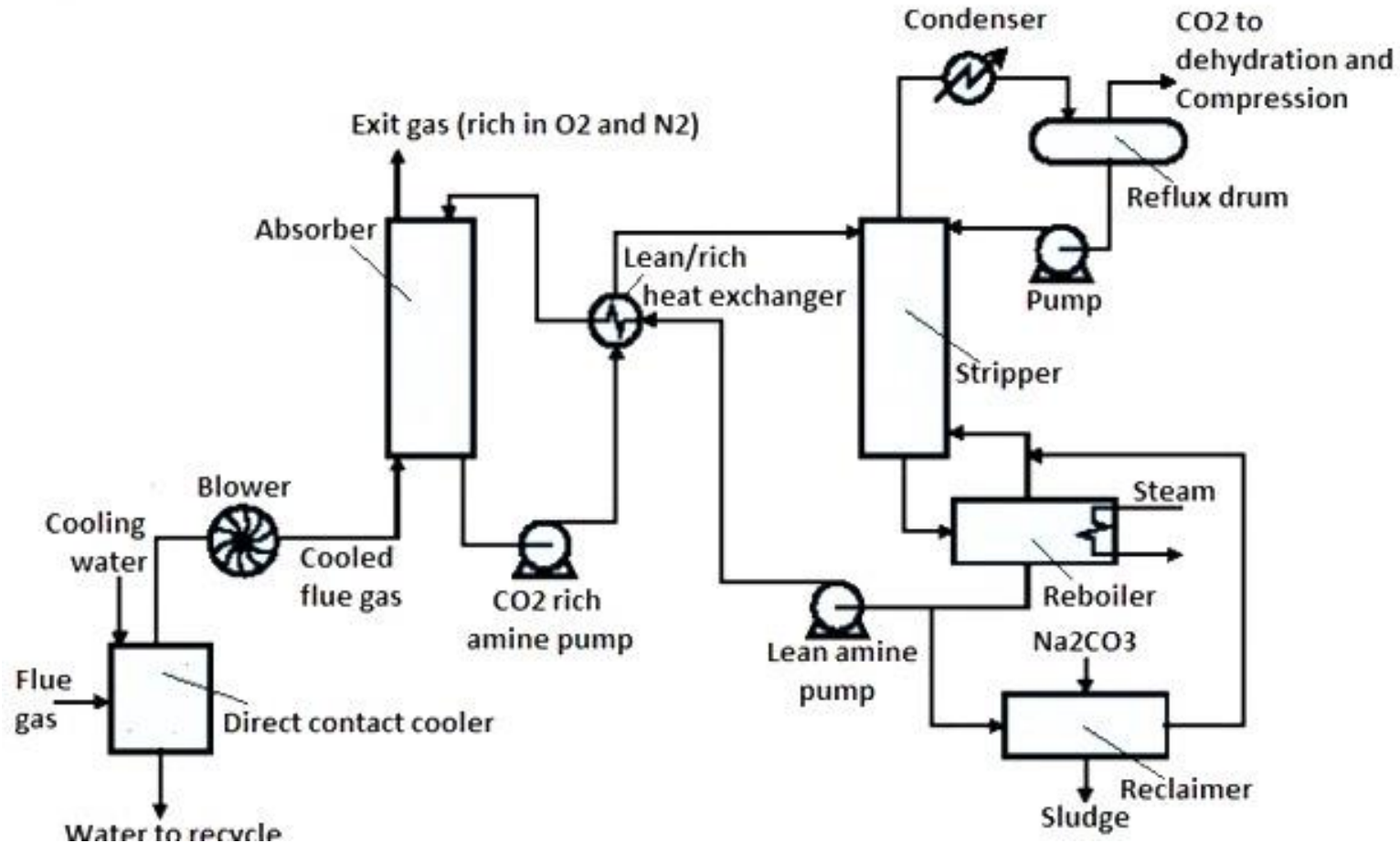
# Simplified Process Diagram

Example of Application to Power Plant Flue Gas (Post-combustion)



# Flow Sheet of CCS

Typical flowsheet of chemical absorption process for CO<sub>2</sub> recovery from flue gas



# Sample Video about CCS

- <https://www.youtube.com/watch?v=EyPI20h9kx0>

- Video by Shell:

<https://www.youtube.com/watch?v=3aIWojhj7Xo>

- Video from GE

<https://www.ge.com/gas-power/future-of-energy/carbon-capture-storage>



# Costs: CAPEX and OPEX

- CAPEX cost for the installation of the system
- OPEX costs in terms of chemicals used, water and power to power the absorber pumps, blowers, cooling towers, steam generation and the CO<sub>2</sub> compressors.



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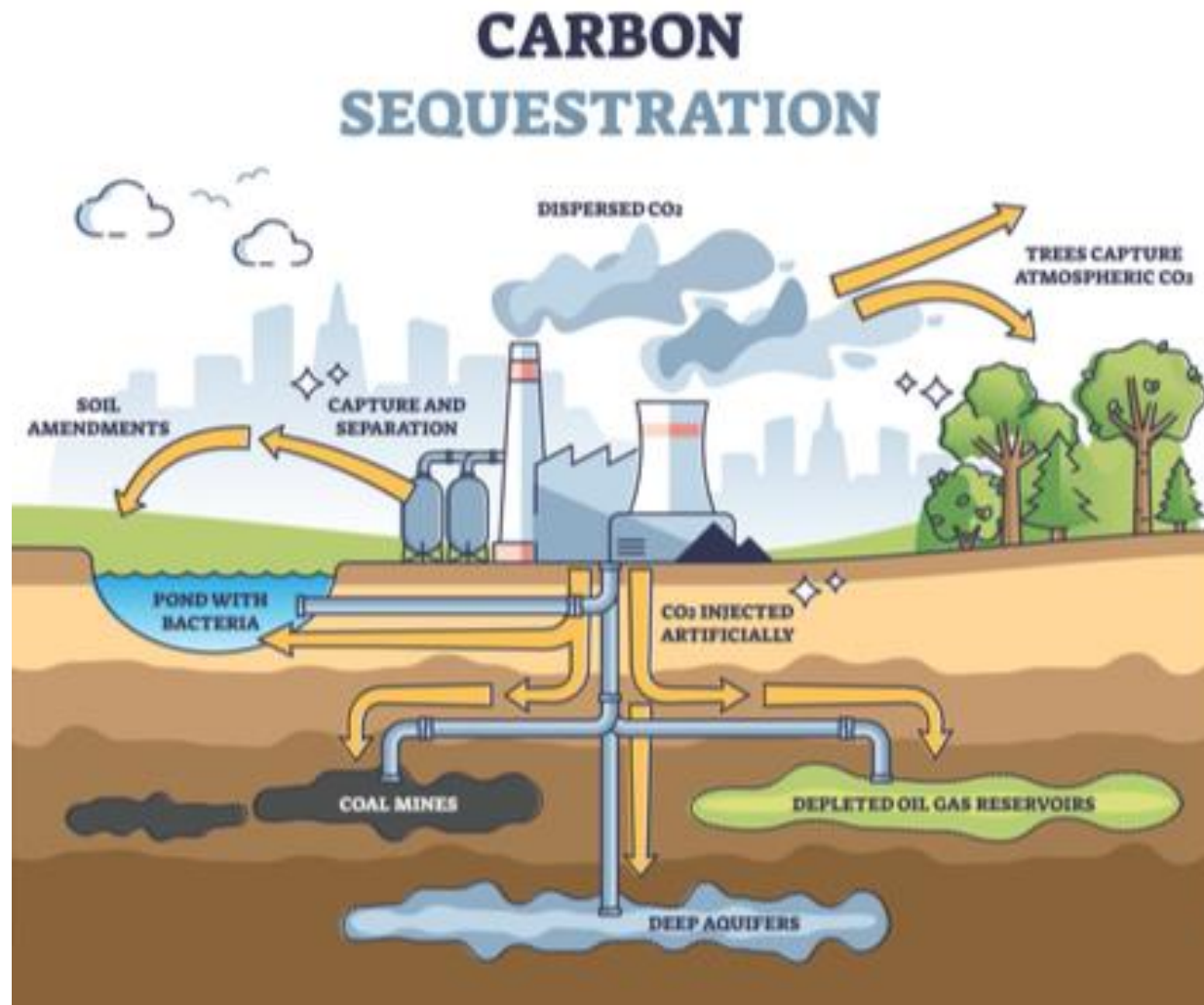
# Uses of CO<sub>2</sub>

- CO<sub>2</sub> is a non-polar acidic gas
- It can be used to decrease the pH or neutralize alkaline waters
- If CO<sub>2</sub> is reduced to CO it can be used a monomer building block
- Labs around the world are trying to synthesize bio-plastics out of CO<sub>2</sub>
- Plants/Trees turn CO<sub>2</sub> and Water into Lignin and Cellulose
- Farms can use CO<sub>2</sub> as a fertilizer/agro-stimulant in green houses
- CO<sub>2</sub> is used in soft drinks
- CO<sub>2</sub> is injected at a low dose in aquariums to feed the plants/algae
- CO<sub>2</sub> gas can be used to enhance oil recovery by pressurising oil formations

# Capture of CO<sub>2</sub> Using Micro-Organisms

- <https://www.theguardian.com/environment/2023/apr/19/volcanic-microbe-eats-co2-astonishingly-quickly-say-scientists>

# CO2 Sequestration Diagram



# Sequestration Geological Locations

- Sedimentary rocks are suitable for the storage of CO<sub>2</sub> due to their frequent high values of porosity and permeability.
- Depleted oil and gas reservoirs, unmined coal beds and saline geological formations are also suitable.
- Basalt formations typically have a composition of 45–52 wt% SiO<sub>2</sub>, 2–5 wt% total alkalis, 0.5–2.0 wt% TiO<sub>2</sub>, 5–14 wt% FeO and 14 wt% or more Al<sub>2</sub>O<sub>3</sub>. Contents of CaO are commonly near 10 wt%, those of MgO commonly in the range 5 to 12 wt%. The latter reacts with CO<sub>2</sub> to form carbonates.

# Pioneers

- GE Power/ALSTOM
- Shell
- ARAMCO
- INEOS
- Linde Engineering
- Scrubbing Agents from BASF
- Occidental Petroleum

# Real Life CCS Projects

- China's largest carbon capture and storage plant at Guohua Jinjie coal power station was completed in January 2021. The project is expected to prevent 150,000 tons of carbon dioxide emission annually at a 90% capture rate.
- In 2020, Norway announced "Longship" ("Langskip" in Norwegian). This 2,7 billion CCS project will capture and store the carbon emissions of Norcem's cement factory in Brevik.
- After the success of their pilot plant operation in November 2011, the Abu Dhabi National Oil Company and Abu Dhabi Future Energy Company moved to create the first commercial CCS facility in the iron and steel industry.<sup>[55]</sup> CO<sub>2</sub> is a byproduct of the iron making process. It is transported via a 50 km pipeline to Abu Dhabi National Oil Company oil reserves for EOR. The facility's capacity is 800,000 tonnes per year. As of 2013, more than 40% of gas emitted by the crude oil production process is recovered within the oil fields for EOR.

# References

- [https://en.wikipedia.org/wiki/List\\_of\\_carbon\\_capture\\_and\\_storage\\_projects](https://en.wikipedia.org/wiki/List_of_carbon_capture_and_storage_projects)
- [https://en.wikipedia.org/wiki/Carbon\\_capture\\_and\\_storage](https://en.wikipedia.org/wiki/Carbon_capture_and_storage)
- <https://www.ge.com/gas-power/future-of-energy/carbon-capture-storage>



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